Name:	e: E	Band:	Date:		
	PSS: Neutralization Reactions				
	Write the net ionic reaction for each of the follow aqueous solutions. a. Carbon acid and strontium hydroxide	ring reactions.	Assume all reactants are		
	b. Perchloric acid and sodium hydroxide				
	c. Hydrobromic acid and barium hydroxide				
	d. Zinc hydroxide and hydrochloric acid				
	e. Aluminum hydroxide and permanganic ad	cid			

2. Identify the formula of the acid and base that formed each of the following salts. Indicate if the acid or base is strong or weak by writing SA, SB, WA, WB in parentheses next to the names of the acid and base.

Salt	Acid	Base
Zinc nitrate		
Aluminum hypochlorite		
Sodium chloride		
Calcium phosphate		
Ammonium oxalate		